

1. (SB 2) Energy lost by the masses = Energy gained by water

$$2(1.50 \text{ kg})(9.8 \text{ ms}^{-2})(3.00 \text{ m}) = (0.2 \text{ kg}) c \Delta T$$

$$c = \text{specific heat of water} = 4186 \text{ J/kg} \cdot ^\circ\text{C}$$

$$\therefore \Delta T = 0.105^\circ\text{C} \#$$

2. (SB 5) By conservation of energy,

$$(1.50 \text{ kg}) c_{\text{iron}} (600^\circ\text{C} - T_f) = (20.0 \text{ kg}) c_{\text{water}} (T_f - 25.0^\circ\text{C})$$

$$\Rightarrow T_f = 29.6^\circ\text{C} \#$$

3. (SB 12)
- $m = \text{mass of ice} = 40.0 \text{ g} = 0.04 \text{ kg}$

Ice ( $-10^\circ\text{C}$ )Ice ( $0^\circ\text{C}$ )Water ( $0^\circ\text{C}$ )Water ( $100^\circ\text{C}$ )Steam ( $100^\circ\text{C}$ )Steam ( $110^\circ\text{C}$ )

$$Q_1 = m c_{\text{ice}} [0^\circ\text{C} - (-10^\circ\text{C})]$$

$$Q_2 = m L_f$$

$$Q_3 = m c_{\text{water}} (100^\circ\text{C} - 0^\circ\text{C})$$

$$Q_4 = m L_v$$

$$Q_5 = m c_{\text{steam}} (110^\circ\text{C} - 100^\circ\text{C})$$

$$\begin{aligned} \text{Total energy needed} &= Q_1 + Q_2 + Q_3 + Q_4 + Q_5 \\ &= 1.22 \times 10^5 \text{ J} \# \end{aligned}$$

4. (SB 16) Conservation of energy

$$(m_{\text{Cu}} c_{\text{Cu}} + m_{\text{water}} c_{\text{water}})(T_f - T_i) = m_{\text{steam}} [L_v + c_{\text{water}} (T_f - 100^\circ\text{C})]$$

$$T_f = 50^\circ\text{C}, T_i = 20^\circ\text{C} \Rightarrow m_{\text{steam}} = 12.9 \text{ g} \#$$

5. (SB 22) (a)
- $W = \text{area under curve}$

$$= 12 \times 10^6 \text{ J} \#$$

$$(b) \text{ Work done by the fluid} = -12 \times 10^6 \text{ J} \#$$

6. (SB 28) (a)
- $W = P(V_f - V_i) = -567 \text{ J} \#$

$$(b) \Delta E_{\text{int}} = Q - W$$

$$= (-400 \text{ J}) - (-567 \text{ J})$$

$$= 167 \text{ J} \#$$

7. (SB 34) For isothermal expansion,
- $W = nRT \ln \frac{V_f}{V_i}$

$$\text{At the final state, } P_f V_f = nRT$$

$$W = 3000 \text{ J}$$

$$P_f = 1 \text{ atm}$$

$$V_f = 25.0 \text{ L}$$

$$n = 1 \text{ mole}$$

Solving for  $T$  and  $V_i$  gives:

$$T = 305 \text{ K} \#$$

$$V_i = 0.00765 \text{ m}^3 \#$$

8. (SB 36) (a)  $\Delta V_{Al} = 3 \alpha_{Al} V_i \Delta T$

$$V_i = \frac{m}{\rho_{Al}} \quad (\rho_{Al} = \text{density of Aluminum})$$

$$\therefore W = P \Delta V_{Al} \quad (P = \text{atmospheric pressure})$$

$$= P \frac{3 \alpha_{Al} m \Delta T}{\rho_{Al}}$$

$$= 0.0486 \text{ J}_\#$$

(b)  $Q = mc_{Al} \Delta T = 16.2 \text{ kJ}_\#$

(c)  $\Delta E_{int} = Q - W \approx 16.2 \text{ kJ}_\#$  (W is negligible)

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9. (SB 61) (a)  $W = \text{area enclosed by the curve}$

$$= 4 P_i V_i \text{ }_\#$$

(b)  $\Delta E_{int} = Q - W = 0$  ( $\because$  initial state = final state)

$$Q = W = 4 P_i V_i \text{ }_\#$$

(c)  $W = 4 P_i V_i$  (from (a))

$$= 4 n R T_i$$

$$= 9.08 \text{ kJ}_\#$$

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10. (SB 25)  $\because P = \text{constant}, \therefore W = P(V_f - V_i)$

But.  $PV_i = nRT_i$

$$PV_f = nRT_f$$

$$\therefore W = P \left( \frac{nR}{P} \right) (T_f - T_i)$$

$$= nR (T_f - T_i)$$

$$= 466 \text{ J}_\#$$